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Diverse volcanism in southeastern Guatemala: The role of crustal contamination

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ABSTRACT

In the Central American arc, southeastern Guatemala hosts the most diverse volcanism. Large stratovolcanoes at the volcanic front (VF) form as a result of subduction of the oceanic Cocos plate beneath the continental Caribbean plate. Behind the volcanic front (BVF) volcanism, however, has undergone a fundamental change in eruptive style during the Quaternary from older, polygenetic central volcanism to younger, monogenetic cinder cone volcanism. Magmas that traverse the 40–45-km-thick crust in southeastern Guatemala are highly susceptible to crustal contamination. Consequently, mineral chemical data, whole-rock oxygen isotope, and light element geochemistry are used to investigate the relationship between edifice type and the magnitude of crustal contamination.

The lack of systematic variation between compositions of phenocryst phases and host rocks strongly suggests that open system processes were operating. Moreover, phenocryst core compositions are generally out of equilibrium with host rock compositions. Olivine from BVF cinder cones deviate only slightly from the equilibrium line in comparison to the older behind the volcanic front (OBVF) central volcanoes and VF stratovolcanoes, suggesting less assimilation of crustal lithologies. Steep arrays on the $\delta^{18}\text{O-SiO}_2$ diagram cannot be explained by crystal fractionation and favor the incorporation of ^{18}O -enriched crustal rocks. Higher $\delta^{18}\text{O}$ values in the OBVF central volcanoes and VF stratovolcanoes support the idea that larger, shallow magma bodies experienced greater amounts of crustal contamination. Regional extension in the Ipala Graben of southeastern Guatemala likely promoted short residence times in crustal reservoirs and small degrees of crustal assimilation for the BVF cinder cone magmas.

Keywords: Arc magmas, geochemistry, stable isotopes, crustal assimilation.

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INTRODUCTION

Magmatic systems can operate in an open or closed manner. Closed magmatic systems allow the exchange of heat with the system's surroundings, but not the exchange of mass. Open magmatic systems can exchange both energy and mass with their surroundings. Different volcanic edifice types in continental arcs result from enigmatic subvolcanic processes. In this paper, we try to ascertain whether open or closed systems processes predominated underneath arc volcanoes in southeastern Guatemala, using select geochemical data. For example, open system behavior, such as crustal assimilation, can change the chemical composition of a magma, including its critical volatile content. Volatile content drives explosive volcanic eruptions, which holds significant implications for the type of volcanic hazards to expect to be associated with the arc volcanoes.

The extent to which crustal contributions modify primary arc magmas remains a controversial issue amongst petrologists. Studies have convincingly demonstrated that crustal inputs play a significant role in the generation of evolved, high-SiO₂ continental arc lavas (Hildreth and Moorbath, 1988; Davidson et al., 1990; Feeley and Sharp, 1995). Greater uncertainty surrounds the contribution of the continental crust in less evolved basaltic arc lavas. In a single arc segment of northern Honshu, Japan, isotopic data for basalts vary systematically with changes in crustal lithology (Kersting et al., 1996). Apparently, even thin crustal lithosphere can modify the compositions of primary melts as they ascend from the mantle.

Although erupted through continental crust, the vast majority of Central American lavas show little evidence for pronounced crustal contamination, based on radiogenic isotopes (Walker et al., 1995; Feigenson et al., 2004). The relatively thin and young continental crust of southern Central America possesses distinct radiogenic isotopic characteristics compared to crust of most other continental margins. Consequently, if crustal assimilation occurs, it does not greatly influence the radiogenic isotopic compositions of the arc magma. The thickness of the continental crust in Central America peaks in western Guatemala at 48 km, from a minimum in Nicaragua of ~32 km (Carr, 1984). Crustal thicknesses decline in central and eastern Guatemala to 45 and 40 km, respectively (Carr et al., 1990). Therefore, Guatemalan lavas hold the most potential for registering crustal geochemical signatures. An unusual trend toward lower ²⁰⁶Pb/²⁰⁴Pb and more radiogenic ²⁰⁷Pb/²⁰⁴Pb and ²⁰⁸Pb/²⁰⁴Pb in southeastern Guatemala cinder cones could result from assimilation of granulitic crust (Walker et al., 1995). Southeastern Guatemala also hosts the most diverse arc volcanism in all of Central America and, consequently, warrants a detailed investigation of the role played by crustal contamination in creating diversity in magma compositions and edifice style.

GEOLOGICAL CONTEXT

Prominent stratovolcanoes or composite cones form a distinct line of volcanoes called the volcanic front (VF) in Guatemala

(Fig. 1). The VF forms as a direct manifestation of subduction of the oceanic Cocos plate beneath the continental Caribbean plate. Common behind the volcanic front (BVF) volcanism adopts contrasting styles in southeastern Guatemala with time. Small composite cones were the site of polygenetic eruptions throughout the late Pliocene and are referred to as older behind the volcanic front (OBVF) central volcanoes. Ubiquitous cinder cones were erupted in proximity to the extensional Ipala Graben (Fig. 1) and locally developed on the flanks of the larger central volcanoes. These monogenetic volcanoes were accordingly assigned Quaternary ages (Walker, 1981). Shield volcanoes such as Las Viboras and calderas such as Retana and Ayarza were also constructed behind the front during the Quaternary (Fig. 1). Overall, however, BVF volcanism in southeastern Guatemala has undergone a fundamental change in eruptive style from older, polygenetic, central volcanism to younger, diffuse, monogenetic, cinder cone volcanism.

Conflicting ideas currently exist concerning the degree of crustal contamination expected to accompany different styles of arc volcanism. A specific plumbing system underlying one of the main edifice types in Guatemala may favor crustal assimilation over the others. In Mexico, as in Guatemala, the cinder cones represent small batches of magma that traverse a separate section of the crust only a single time (Hochstaedter et al., 1996). This eruptive character may make more differentiated members of the cinder cone suites more susceptible to crustal assimilation during their ascent to the surface. In sharp contrast, composite cones may erupt repeatedly through well-established plumbing systems. This feature suggests that older, crustally contaminated lavas may give way to larger volumes of magmas that were shielded in the conduit of the volcano from interacting with crustal rocks. Conversely, crustal contamination might flourish beneath composite cones where magma stalls in large crustal magma reservoirs. In Mexico, magmas erupted from cinder cones seem to experience more crustal contamination (Hochstaedter et al., 1996). In Guatemala, the appropriate data have not existed to test this hypothesis.

Detecting the geochemical signatures of crustal contamination depends not only on the degree of assimilation but also on the compositional contrast between magma and contaminant. Systematic increases in Sr isotopic ratios and decreases in Nd isotopic ratios across the arc in BVF cinder cones of Guatemala reflect either an absolute or apparent increase in crustal contamination (Walker et al., 1995; Feigenson et al., 2004). Absolute increases in crustal contamination associated with cinder cones from the Ipala Graben far behind the VF make geologic sense, if the normal faulting that accompanies extension facilitates assimilation by increasing the surface area of crustal lithologies. Alternatively, the extensional regime of eruption could result in less impedance and more direct access to the surface and, consequently, smaller degrees of contamination. In contrast, assimilation of equal amounts of older, more radiogenic crust in the Ipala Graben could also generate the observed isotopic changes. Low Nd isotopic ratios thought to represent crustal contamination were also measured in mafic lavas from Tegucigalpa in Honduras (Patino et al., 1997).

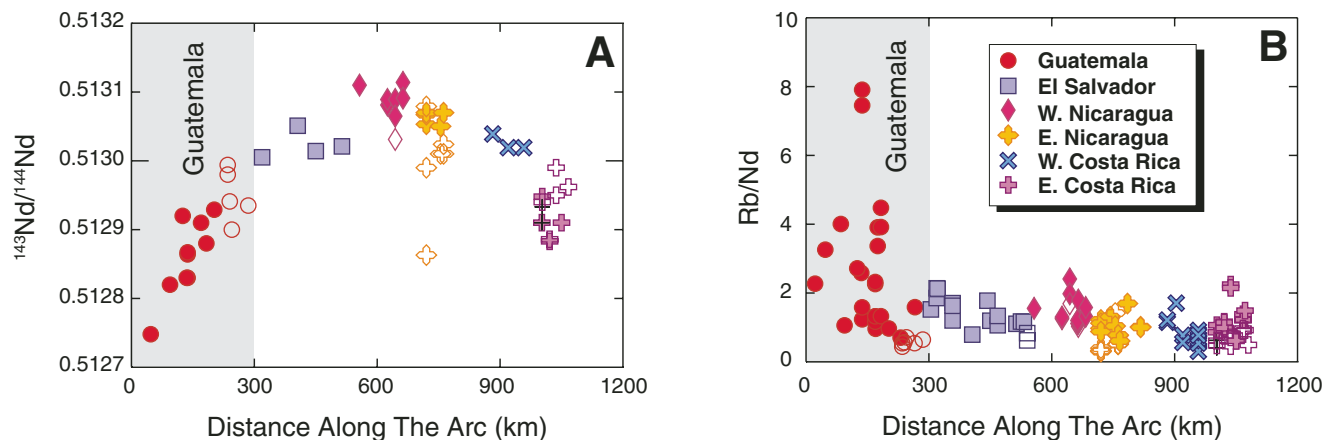


Figure 2. Variations in mafic samples along the Central American arc in geochemical parameters that indicate crustal contamination is most likely in southeastern Guatemala. (A) Low $^{143}\text{Nd}/^{144}\text{Nd}$ occurs in Guatemala (circles). Behind the volcanic front (BVF) volcanism is shown by open circles. Crustal contamination or anomalous mantle may explain the complementary low $^{143}\text{Nd}/^{144}\text{Nd}$ in Costa Rica (crosses). (B) Guatemalan lavas possess high Rb/Nd ratios compared to the rest of the Central American arc.

and phenocryst phases were measured with a Cameca SX-50 electron microprobe at the University of Chicago, calibrated with natural and synthetic standards. The program audit profile (PAP) correction routine was utilized for all analyses. Typical analytical conditions adopted were a 15 kV accelerating voltage and a 25 nA beam current. Most analyses were run with a focused beam and a beam width of 1–2 μm . Tables 1–3 report representative major element compositions and mineral formulae of olivine, pyroxene, and plagioclase.

Significant crustal inputs of the light elements (i.e., Li, Be, and B) to arc magmas seem especially probable where granitoid rocks serve as the contaminant. Consequently, the light elements were measured on a subset of 23 lavas from southeast Guatemala and two possible crustal contaminants by direct current plasma atomic emission spectrometry (DCP-AES) using an ARL model SpectraSpan (SS) 7 DCP at the University of South Florida. Three standards were run for B (RR-1, NBS-688, and QDF-1), and two standards (STM-1 and QLO-1) were analyzed for Li and Be to

TABLE 1. REPRESENTATIVE OLIVINE MINERAL CHEMICAL DATA

| Sample Group | GIP.1Pc* OBVF | GIP.1Pr OBVF | GIP.1GM OBVF | GCC837.Pc BVF | GCC837.Pr BVF | GCC837.GM BVF | G201.GM BVF | T302.Pc VF | T302.Pr VF |
|----------------------------|------------------|-----------------|-----------------|------------------|------------------|------------------|----------------|---------------|---------------|
| SiO ₂ | 37.08 | 34.19 | 34.79 | 39.62 | 38.14 | 37.06 | 36.53 | 39.07 | 37.96 |
| MgO | 34.35 | 22.35 | 23.59 | 41.12 | 36.18 | 31.15 | 29.32 | 41.51 | 35.32 |
| FeO | 27.16 | 41.76 | 40.34 | 18.53 | 24.34 | 30.49 | 32.82 | 18.28 | 25.24 |
| MnO | 0.42 | 0.89 | 0.78 | 0.22 | 0.48 | 0.63 | 0.73 | 0.23 | 0.42 |
| CaO | 0.19 | 0.3 | 0.29 | 0.15 | 0.2 | 0.35 | 0.33 | 0.13 | 0.16 |
| NiO | 0.05 | 0.05 | 0 | 0.02 | 0.03 | 0.05 | 0.08 | 0.07 | 0.06 |
| Total | 99.25 | 99.54 | 99.79 | 99.66 | 99.37 | 99.73 | 99.81 | 99.29 | 99.16 |
| Formula based on 4 oxygens | | | | | | | | | |
| Si | 0.998 | 0.993 | 0.998 | 1.013 | 1.009 | 1.008 | 1.006 | 1.004 | 1.011 |
| Mg | 1.378 | 0.968 | 1.009 | 1.568 | 1.427 | 1.264 | 1.204 | 1.59 | 1.402 |
| Fe | 0.611 | 1.014 | 0.968 | 0.396 | 0.538 | 0.694 | 0.756 | 0.393 | 0.562 |
| Mn | 0.01 | 0.022 | 0.019 | 0.005 | 0.011 | 0.015 | 0.017 | 0.005 | 0.009 |
| Ca | 0.005 | 0.009 | 0.009 | 0.004 | 0.006 | 0.01 | 0.01 | 0.004 | 0.005 |
| Ni | 0.001 | 0.001 | 0 | 0 | 0.001 | 0.001 | 0.002 | 0.001 | 0.001 |
| mol% end members | | | | | | | | | |
| Fo | 69.3 | 48.8 | 51.0 | 79.8 | 72.6 | 64.6 | 61.4 | 80.2 | 71.4 |
| Fa | 30.7 | 51.2 | 49.0 | 20.2 | 27.4 | 35.4 | 38.5 | 19.8 | 28.6 |

Note: BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

*The prefix of the sample designations stands for Guatemala and the particular volcano, whereas the suffix is composite of texture (eg. P—phenocryst, GM—groundmass) and analysis location (eg. c—core or r—rim).

assess precision and accuracy. Results of other studies quote errors within 10% (relative) of accepted values for both B and Be (Hochstaedter et al., 1996). In this study, precision for B was closer to 20% at the 1 ppm level, but improved to ~1% at the 10 ppm level. Errors on the Be standards were 11 and 16% at the 1 ppm and 10 ppm level, respectively. Li analyses had errors between 0.06% and 4.1%. Table 4 presents the new Li, Be, and B data.

Oxygen-isotopic ratios were determined on ~12 mg samples of 27 separate whole-rock powders at the Stable Isotope Laboratory at Southern Methodist University in Dallas. Oxygen was extracted from powders following reaction with ClF_3 . Typical analytical precision on whole-rock $\delta^{18}\text{O}$ values approximated $\pm 0.2\%$ based on standards and repeat analyses. Of the whole-rock powders analyzed, two are crustal rocks, seven are lavas from VF stratovolcanoes, nine are from OBVF central volcanoes, and nine are from BVF cinder cones. Table 4 also reports the ana-

lytical results in the standard per mil (‰) notation as deviations relative to the standard mean ocean water (SMOW) standard.

RESULTS

Mineral Chemical Data

Olivine occurs as a phenocryst and groundmass phase in basalts and basaltic andesites in southeast Guatemala (Table 1). Normal zoning displayed by backscatter images was confirmed by electron microprobe analyses. Olivine cores (Fig. 3A; mean $\text{Fo}_{73.1}$) are generally more enriched in Mg relative to rim compositions (Fig. 3B; mean $\text{Fo}_{64.5}$). Groundmass olivine grains (mean $\text{Fo}_{56.2}$) are more Mg-poor than both phenocryst cores and rims (Fig. 3C).

A comparison of olivine compositions at the different edifices shows that olivine cores at VF composite cones (mean

TABLE 2. REPRESENTATIVE PYROXENE MINERAL CHEMICAL DATA

| Sample Group | GJM.2Pc* OBVF | GJM.2Pr OBVF | GJM.2GM OBVF | GCC911.GM BVF | GCC837.Pc BVF | T302.Pc VF | T302.Pr VF | M210.Pc VF | M210.Pr VF |
|----------------------------|------------------|-----------------|-----------------|------------------|------------------|---------------|---------------|---------------|---------------|
| SiO_2 | 52.2 | 51.78 | 49.35 | 52.57 | 48.36 | 50.76 | 51.9 | 51.62 | 51.04 |
| Al_2O_3 | 2.15 | 1.92 | 3.68 | 1.51 | 4.29 | 4.09 | 2.94 | 2.03 | 2.1 |
| Fe_2O_3 | 2.06 | 2.47 | 2.85 | 0 | 3.46 | 1.5 | 0.43 | 0.11 | 0.1 |
| TiO_2 | 0.42 | 0.39 | 1.48 | 0.57 | 1.48 | 0.7 | 0.54 | 0.5 | 0.52 |
| Cr_2O_3 | 0.01 | 0 | 0.09 | 0.18 | 0.06 | 0.03 | 0.03 | 0.01 | 0.01 |
| MgO | 24.67 | 24.74 | 14.47 | 17.06 | 12.96 | 14.89 | 15.22 | 13.79 | 13.09 |
| FeO | 15.92 | 15.27 | 7 | 11.85 | 8.4 | 7.32 | 8.53 | 11.66 | 10.72 |
| MnO | 0.38 | 0.44 | 0.24 | 0.4 | 0.29 | 0.23 | 0.29 | 0.4 | 0.4 |
| CaO | 1.84 | 1.75 | 19.87 | 15.06 | 20.02 | 20 | 19.73 | 18.57 | 19.63 |
| Na_2O | 0.03 | 0.03 | 0.37 | 0.2 | 0.38 | 0.34 | 0.27 | 0.37 | 0.41 |
| Total | 99.68 | 98.79 | 99.4 | 99.4 | 99.7 | 99.86 | 99.88 | 99.06 | 98.02 |
| Formula based on 6 oxygens | | | | | | | | | |
| Si | 1.915 | 1.915 | 1.85 | 1.96 | 1.826 | 1.882 | 1.925 | 1.953 | 1.951 |
| Al | 0.085 | 0.084 | 0.15 | 0.04 | 0.174 | 0.118 | 0.075 | 0.047 | 0.049 |
| Fe^{3+} | 0 | 0.001 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| Al | 0.008 | 0 | 0.013 | 0.026 | 0.017 | 0.061 | 0.054 | 0.044 | 0.046 |
| Fe^{3+} | 0.057 | 0.068 | 0.08 | 0 | 0.098 | 0.42 | 0.012 | 0.003 | 0.003 |
| Ti | 0.012 | 0.011 | 0.042 | 0.016 | 0.042 | 0.2 | 0.015 | 0.014 | 0.015 |
| Cr | 0 | 0 | 0.003 | 0.005 | 0.002 | 0.001 | 0.001 | 0 | 0 |
| Mg | 0.923 | 0.921 | 0.809 | 0.948 | 0.729 | 0.823 | 0.841 | 0.778 | 0.746 |
| Fe^{2+} | 0 | 0 | 0.053 | 0.005 | 0.112 | 0.053 | 0.077 | 0.161 | 0.19 |
| Mn | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| Mg | 0.426 | 0.443 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| Fe^{2+} | 0.488 | 0.472 | 0.166 | 0.365 | 0.153 | 0.174 | 0.188 | 0.208 | 0.153 |
| Mn | 0.012 | 0.014 | 0.008 | 0.013 | 0.009 | 0.007 | 0.009 | 0.013 | 0.013 |
| Ca | 0.072 | 0.069 | 0.798 | 0.602 | 0.81 | 0.795 | 0.784 | 0.753 | 0.804 |
| Na | 0.002 | 0.002 | 0.027 | 0.014 | 0.028 | 0.24 | 0.019 | 0.027 | 0.03 |
| mol% end members | | | | | | | | | |
| Wo | 3.8 | 3.6 | 43.7 | 31.3 | 44.9 | 43.1 | 41.5 | 39.6 | 42.5 |
| En | 70.7 | 71.6 | 44.3 | 49.4 | 40.4 | 44.6 | 44.5 | 41.0 | 39.4 |
| Fs | 25.6 | 24.8 | 12.0 | 19.3 | 14.7 | 12.3 | 14.0 | 19.4 | 18.1 |

Note: BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

*The prefix of the sample designations stands for Guatemala and the particular volcano, whereas the suffix is composite of texture (eg. P—phenocryst, GM—groundmass) and analysis location (eg. c—core or r—rim).

TABLE 3. REPRESENTATIVE PLAGIOCLASE MINERAL CHEMICAL DATA

| Sample Group | GJM.2Pc OBVF | GJM.2Pr OBVF | GJM.2GM OBVF | GCH.1GM OBVF | GCC837.Pc BVF | GCC837.Pr BVF | GCC837.GM BVF | T302.Pc VF | T302.Pr VF |
|--------------------------------|-----------------|-----------------|-----------------|-----------------|------------------|------------------|------------------|---------------|---------------|
| SiO ₂ | 52.86 | 51.22 | 54.14 | 52.89 | 46.63 | 49.71 | 49.73 | 45.08 | 49.5 |
| Al ₂ O ₃ | 28.78 | 30.24 | 27.7 | 29.08 | 33.48 | 30.69 | 31.37 | 34.35 | 31.3 |
| Na ₂ O | 4.47 | 3.48 | 5.08 | 4.37 | 1.59 | 3.07 | 2.85 | 1.27 | 3.07 |
| CaO | 12.18 | 13.96 | 11.02 | 12.07 | 17.43 | 14.77 | 15.13 | 17.64 | 14.79 |
| K ₂ O | 0.25 | 0.18 | 0.34 | 0.34 | 0.05 | 0.15 | 0.13 | 0 | 0.06 |
| BaO | 0 | 0 | 0 | 0.04 | 0 | 0 | 0 | 0 | 0 |
| Total | 98.54 | 99.08 | 98.28 | 98.79 | 99.18 | 98.39 | 99.21 | 98.34 | 98.72 |
| Formula based on 32 oxygens | | | | | | | | | |
| Si | 9.713 | 9.401 | 9.945 | 9.694 | 8.64 | 9.217 | 9.147 | 8.435 | 9.145 |
| Al | 6.233 | 6.542 | 5.998 | 6.282 | 7.312 | 6.707 | 6.801 | 7.576 | 6.816 |
| Na | 1.593 | 1.238 | 1.809 | 1.553 | 0.571 | 1.104 | 1.016 | 0.461 | 1.1 |
| Ca | 2.398 | 2.745 | 2.169 | 2.37 | 3.461 | 2.935 | 2.982 | 3.537 | 2.928 |
| K | 0.059 | 0.042 | 0.08 | 0.08 | 0.012 | 0.035 | 0.031 | 0 | 0.014 |
| Ba | 0 | 0 | 0 | 0.003 | 0 | 0 | 0 | 0 | 0 |
| mol% end members | | | | | | | | | |
| Ab | 39.3 | 30.8 | 44.6 | 38.8 | 14.1 | 27.1 | 25.2 | 11.5 | 27.2 |
| An | 59.2 | 68.2 | 53.4 | 59.2 | 85.6 | 72.0 | 74.0 | 88.5 | 72.4 |

Note: BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

*The prefix of the sample designations stands for Guatemala and the particular volcano, whereas the suffix is composite of texture (eg. P—phenocryst, GM—groundmass) and analysis location (eg. c—core or r—rim).

TABLE 4. OXYGEN AND Sr ISOTOPIC DATA ALONG WITH SELECT TRACE ELEMENTS

| Sample | Group | SiO ₂ (wt%) | δ ¹⁸ O (‰) | ⁸⁷ Sr/ ⁸⁶ Sr [†] | B (ppm) | Be (ppm) | Li (ppm) | La (ppm) | Pb (ppm) | Th (ppm) | Cs (ppm) | U (ppm) | Rb (ppm) | Sb (ppm) | Ba (ppm) |
|---------|-------|---------------------------|--------------------------|---|------------|-------------|-------------|-------------|-------------|-------------|-------------|------------|-------------|-------------|-------------|
| G2 | Crust | 59.65 | 3.71 | 0.7066 | 24.1 | 1.13 | 21.7 | 21.21 | 14.88 | 8.43 | 0.39 | 1.26 | 1.33 | 66.02 | 497.3 |
| M2 | Crust | 81.19 | 14.27 | 0.74277 | 22.0 | 0.45 | 1.9 | 34.1 | 5.76 | 6.78 | 5.612 | 1.773 | | | 1377 |
| GUC-201 | BVF | 50.36 | 6.47 | 0.703547 | | | | 12.53 | 5.17 | 1.25 | 0.53 | 0.53 | 14.12 | 0.17 | 404.8 |
| GUC-911 | BVF | 53.71 | 7.74 | 0.704054 | 9.9 | 1.29 | 9.0 | 28.53 | 7.14 | 1.19 | 0.6 | 0.5 | 25.12 | 0.1 | 1039 |
| GUC-837 | BVF | 49.01 | 6.61 | 0.703975 | 2.3 | 0.87 | 7.8 | 14.05 | 5.25 | 0.96 | 0.39 | 0.41 | 17.41 | 0.09 | 344.4 |
| GUC-839 | BVF | 48.32 | 6.7 | | | | | 11.73 | 3.58 | 0.82 | 0.16 | 0.34 | 7.94 | 0.03 | 263.6 |
| GUC-844 | BVF | 50.39 | 6.63 | 0.703791 | 2.7 | 1.07 | 8.4 | 14.21 | 4.9 | 1.48 | 0.27 | 0.64 | 23.31 | 0.05 | 462.2 |
| GUC-702 | BVF | 49.98 | 6.72 | 0.70353 | 2.7 | 1.33 | 9.2 | 15.18 | 4.1 | 1.07 | 0.24 | 0.5 | 13.6 | 0.02 | 402.2 |
| GUC-800 | BVF | 48.76 | 6.16 | | 1.1 | 0.85 | 7.4 | 10 | 2.59 | 0.95 | 0.15 | 0.22 | 7.33 | | 248.3 |
| GUC-25 | BVF | 53.7 | | 0.70381 | 10.0 | 1.66 | 12.3 | 26 | 6.6 | 2.07 | 0.26 | 0.82 | 20.88 | 0.05 | 624.7 |
| GUC-303 | BVF | 49.9 | | 0.70321 | 4.0 | 0.92 | 7.3 | 9 | 2.25 | 0.67 | 0.04 | 0.17 | 6.64 | | 181.4 |
| GUAT-20 | BVF | | 6.67 | | | | | | | | | | | | |
| GUAT-33 | BVF | | 6.98 | | | | | | | | | | | | |
| GUC-835 | OBVF | 50.33 | 6.87 | 0.70415 | | | | 15.13 | 4.85 | 1.13 | 0.37 | 0.53 | 17.65 | 0.05 | 500.5 |
| GSU-01 | OBVF | 52.56 | 7.5 | 0.704135 | 8.3 | 0.86 | 9.4 | 12 | 6.71 | 1.89 | 1.09 | 0.69 | 36.77 | 0.17 | 470.6 |
| GSU-04 | OBVF | 55.43 | 7.41 | 0.704101 | 4.5 | 1.24 | 9.2 | 22.88 | 8.91 | 3.1 | 1.41 | 1.22 | 50.99 | 0.13 | 668.2 |
| GIP-01 | OBVF | 52 | 7.14 | 0.703877 | 4.1 | 1.36 | 10.0 | 18.6 | 6.45 | 1.93 | 0.83 | 0.78 | 29.92 | 0.17 | 504.8 |
| GIP-04 | OBVF | 51.04 | 7.12 | 0.703884 | 2.6 | 0.87 | 8.6 | 15.13 | 4.43 | 1.47 | 0.35 | 0.63 | 18.96 | 0.07 | 480.8 |
| GJM-02 | OBVF | 56.67 | 7.86 | | 11.5 | 1.08 | 17.3 | 21.74 | 7.64 | 3.9 | 1.38 | 1.26 | 47.37 | 0.15 | 676.9 |
| GJM-03 | OBVF | 51.73 | 7.85 | | 6.8 | 0.76 | 11.5 | 12.64 | 4.97 | 2.04 | 0.89 | 0.68 | 26.93 | 0.11 | 480.7 |
| GCH-01 | OBVF | 51.15 | 6.9 | | 2.6 | 1.01 | 10.0 | 20.08 | 4.47 | 1.47 | 0.67 | 0.61 | 17.33 | 0.08 | 582.2 |
| GUC-812 | OBVF | 49.91 | 6.51 | | 3.7 | 0.98 | 8.4 | 16.04 | 4.39 | 1 | 0.12 | 0.28 | 9.77 | 0.02 | 354.2 |
| LC-2 | OBVF | 54.75 | | 0.704102 | 8.0 | 1.16 | 10.1 | 23.49 | 7.98 | 2.28 | 0.78 | 0.78 | 38.61 | 0.18 | 865.8 |
| FEL-3 | OBVF | 58.25 | | 0.70409 | 18.1 | 1.23 | 11.4 | 23.35 | 12.1 | 3.01 | 0.73 | 1.04 | 54.22 | 0.3 | 935.9 |
| FVH-4 | OBVF | 64.32 | | 0.704056 | 22.8 | 1.33 | 10.4 | 25.61 | 13.6 | 5.02 | 2.32 | 1.83 | 99.48 | 0.39 | 943 |
| GPA-01 | VF | 49.26 | 6.36 | | | | | 8.2 | 3.26 | 0.96 | 0.41 | 0.42 | 10.26 | 0.02 | 341.2 |
| GAG-21 | VF | 51.4 | 6.59 | | 6.0 | 0.75 | | 10.48 | 5.49 | 1.57 | 0.62 | 0.67 | 19.53 | 0.2 | 519.9 |
| GAG-27 | VF | 55.31 | 6.79 | | | | | 13.19 | 5.07 | 2.21 | 0.6 | 0.99 | 39.69 | 0.25 | 638.1 |
| GAG-10 | VF | 59.1 | 6.78 | | | | | 16.09 | 8.12 | 3.17 | 1.93 | 1.46 | 55.92 | 0.4 | 748.7 |
| TCB-302 | VF | 53.87 | 6.73 | | | | | 5.83 | 3.27 | 0.45 | 0.77 | 0.25 | 9.41 | 0.54 | 314.7 |
| E1 | VF | 51.4 | 7.44 | 0.70389 | 3.1 | 0.62 | 9.2 | 10.15 | 5.51 | 1.47 | 0.58 | 0.58 | 13.53 | | 442.3 |
| GMO-201 | VF | 57.46 | 8.41 | 0.703649 | 22.9 | 0.78 | 9.6 | 12.29 | 8.12 | 3.43 | 2.11 | 1.48 | 46.59 | 0.7 | 669 |
| Qam-13 | VF | 49.81 | | | 7.0 | 0.58 | 9.2 | 8.51 | 4.6 | 1.87 | 0.64 | 0.62 | 13.6 | | 486.6 |

Note: BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

[†]Sr isotopic ratios were obtained at the mass spectrometry facility at Rutgers University. Sr isotopic ratios are normalized to ⁸⁶Sr/⁸⁸Sr of 0.1194 and are reported as measured.

$Fo_{75.3}$; Fig. 3A) and BVF cinder cones (mean $Fo_{76.0}$; Fig. 3A) have slight enrichments in Mg compared to the OBVF central volcanoes (mean $Fo_{71.6}$; Fig. 3A). Moreover, VF stratovolcanoes and BVF cinder cones have more narrow ranges of olivine core compositions (Fig. 3). In contrast, OBVF central volcanoes contain olivine that extends the range in forsterite (Fo) content to below 60 mol% (Fig. 3).

Clinopyroxene occurs as both a phenocryst and groundmass phase in most lavas from southeast Guatemala, except for BVF cinder cones, which only contain clinopyroxene as a groundmass phase. In contrast to olivine, backscatter images and electron microprobe analyses showed little chemical zoning in the

clinopyroxene phenocrysts. Phenocryst cores (Fig. 4A; mean Mg number = 77.0) have nearly identical ranges and means as phenocryst rims (Fig. 4B; mean Mg number = 77.4). Groundmass clinopyroxene grains possess a similar range in Mg number as phenocrysts (Fig. 4C), but have a slightly lower average value (mean Mg number = 74.2). More evolved samples, such as the dacite (Figs. 4A and 4B), display both normal and reversed zoning, though variations in Mg number are relatively small. Clinopyroxene compositions vary little as whole-rock SiO_2 increases.

Clinopyroxene core compositions at VF stratovolcanoes (mean Mg number = 77.0) and OBVF central volcanoes (mean Mg number = 77.0) have similar ranges (Fig. 4A) and an identical mean. Owing to the paucity of clinopyroxene phenocrysts in cinder cone lavas, only one phenocryst core was measured in a BVF cinder cone rock, and its Mg# is 73.3 (Fig. 4A). The range for groundmass clinopyroxene in BVF cinder cones is shifted to lower Mg numbers (62.2–80.8) relative to the phenocrysts in lavas from the other edifice types (Fig. 4). The mean Mg number of the groundmass clinopyroxene at BVF cinder cones (Mg number = 72.7) is lower than the clinopyroxene phenocryst core and rim compositions from the other edifice types, but approximates the mean from the lone phenocryst analysis from a Guatemalan cinder cone (Fig. 4A).

Orthopyroxene occurs in lower modal abundances as phenocryst and groundmass phases than clinopyroxene in most Guatemalan lavas. Groundmass and phenocryst orthopyroxene exhibit a similar range of compositions (Fig. 5). Moreover, orthopyroxene phenocryst cores, rims, and groundmass grains have nearly identical mean Mg numbers (mean Mg number of 71.9, 70.9, and 70.3, respectively). Despite these similar means, the data show both normal and reversely zoned phenocrysts in some of the more evolved magmas. It is noteworthy that Mg-rich cores persist across the compositional spectrum. The lower Mg number of orthopyroxene cores (mean Mg number = 71.9) compared to clinopyroxene (mean Mg number = 77.0) in the lavas from southeast Guatemala may indicate that orthopyroxene occurs later in the crystallization history.

Slight variations in orthopyroxene core composition exist based on the type of volcanic edifice (Fig. 5). Although orthopyroxene phenocrysts from OBVF central volcanoes and VF stratovolcanoes have similar ranges in Mg number (Fig. 5A), orthopyroxene cores in lavas from the former edifice type (mean Mg number = 73.1) have a higher mean Mg number than the latter (mean Mg number = 70.2). No orthopyroxene phenocrysts from BVF cinder cones were analyzed by the electron microprobe.

Plagioclase occurs as a modally abundant phenocryst and groundmass phase in the arc lavas of southeast Guatemala. Both phenocryst cores and rims exhibit a nearly identical, wide range of An contents (Figs. 6A and 6B), yet generally display normal zoning. The mean An values for plagioclase phenocryst cores and rims ($An_{70.5}$ and $An_{61.6}$, respectively) reflects the predominance of normally zoned crystals. However, some more-evolved samples

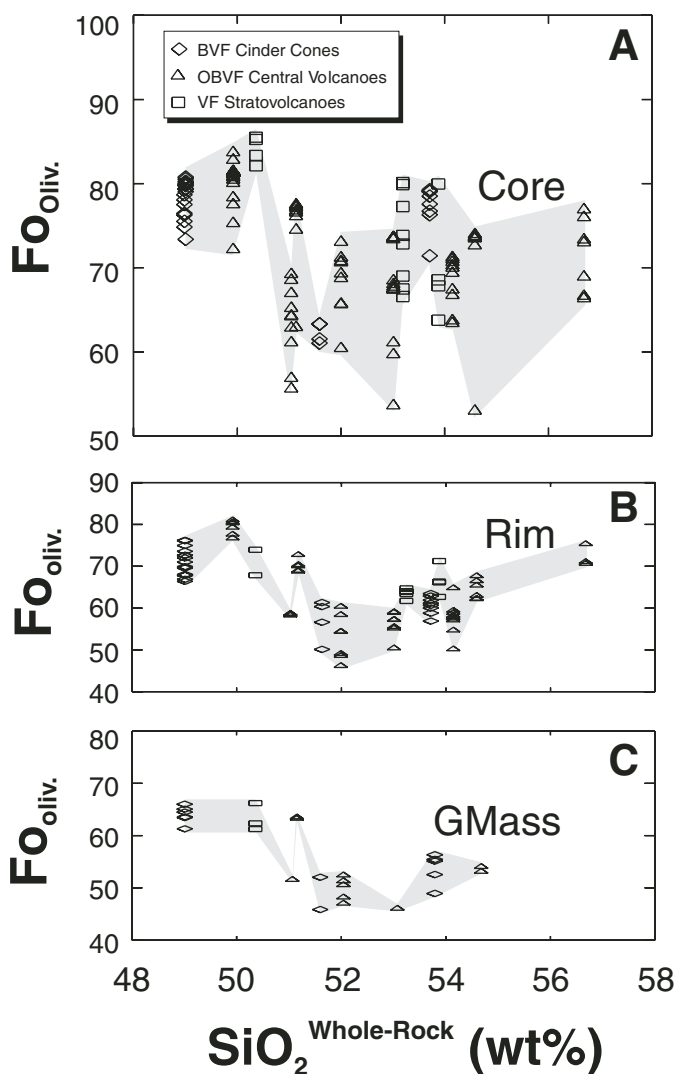


Figure 3. Forsterite (Fo) content variations in olivine against whole rock SiO_2 compositions. (A) Core Fo contents in olivine phenocrysts. (B) Rim Fo contents in olivine phenocrysts. (C) Groundmass (GMass) olivine Fo contents. Shaded compositional envelopes were drawn to highlight that little systematic variation exists between the Fo contents and the host rock chemistry.

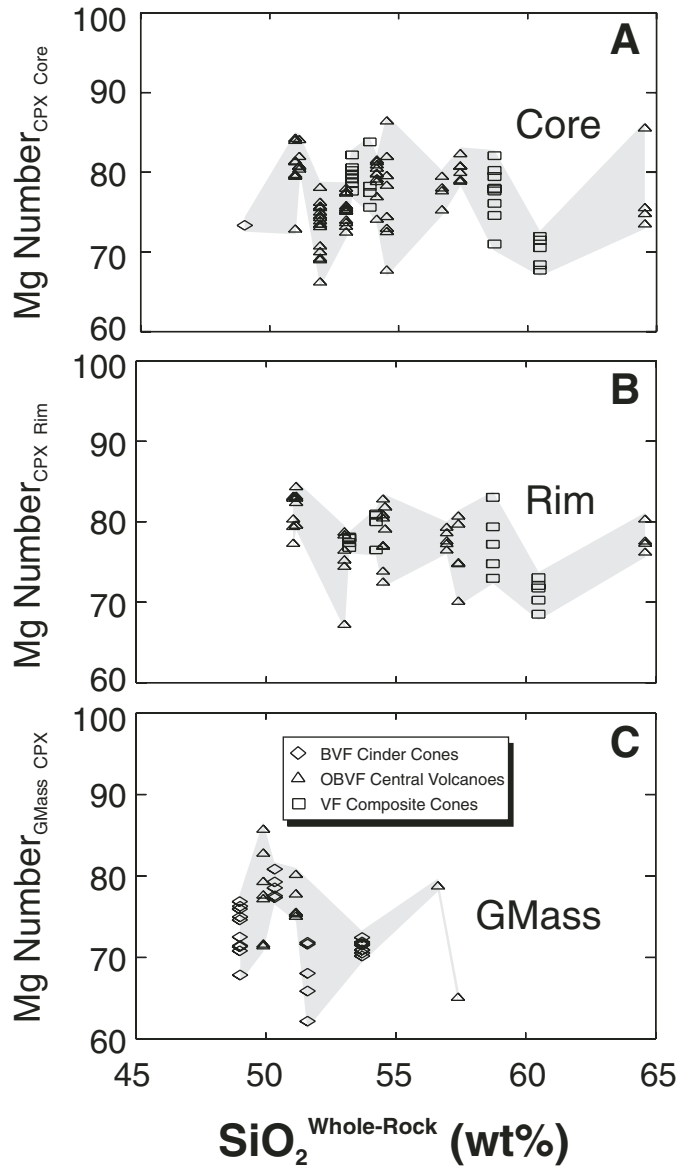


Figure 4. Mg number variations in clinopyroxene grains versus whole-rock SiO_2 compositions. (A) Core Mg number values of clinopyroxene phenocrysts. (B) Rim Mg number values of clinopyroxene phenocrysts. (C) Mg number contents of groundmass (GMass) clinopyroxene grains. Shaded compositional envelopes were drawn to highlight the nonsystematic relationship between Mg number in clinopyroxene versus host rock composition.

do display both normal and reversed zoning. A lower mean An content in groundmass plagioclase (mean $\text{An}_{58.6}$) relative to phenocryst rims (mean $\text{An}_{61.6}$) continues the overall shift in the range of An contents to lower values with inferred crystallization order. An content in plagioclase crystals varies considerably at whole-rock SiO_2 contents <57 wt%, but generally decreases at SiO_2 contents >57 wt% (Figs. 6A and 6B).

Plagioclase crystals of varying composition occur in lavas erupted from the different edifice types (Fig. 6). Plagioclase

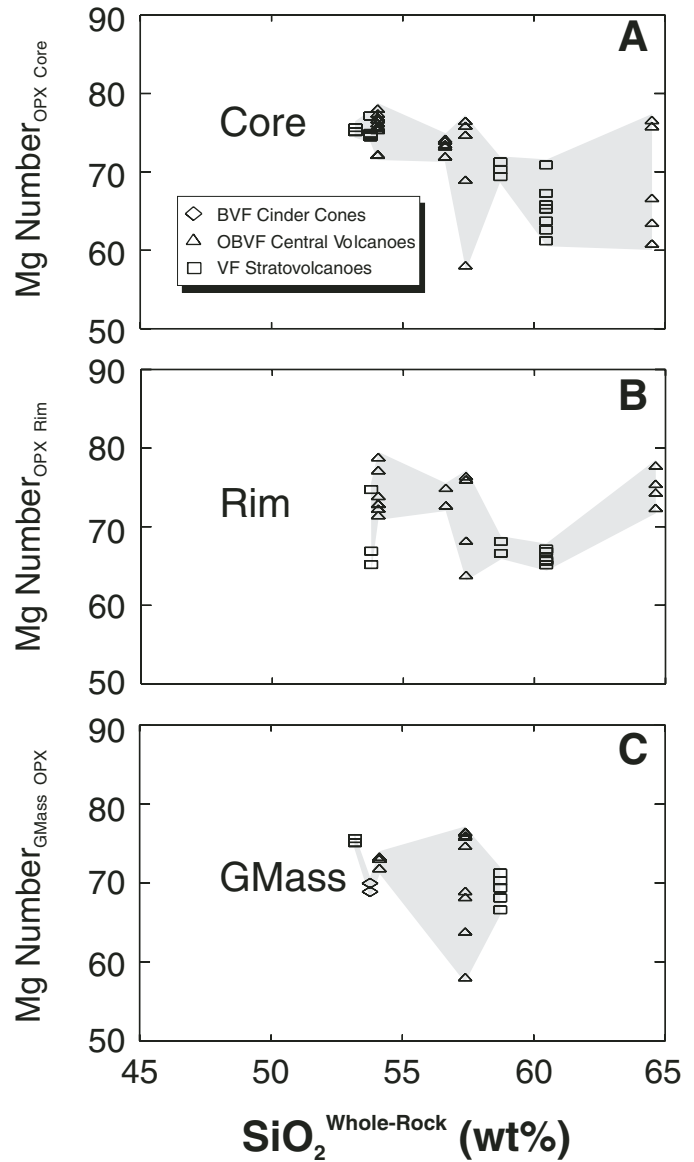


Figure 5. Mg number variations in orthopyroxene grains versus whole-rock SiO_2 compositions. (A) Mg number values for orthopyroxene phenocrysts. (B) Rim Mg number values of orthopyroxene phenocrysts. (C) Mg number values for groundmass (GMass) orthopyroxene. Shaded compositional envelopes were drawn to highlight that Mg number in orthopyroxene remains relatively constant over a range of host rock SiO_2 compositions.

cores from polygenetic VF stratovolcanoes and OBVF central volcanoes exhibit large ranges in An content (Fig. 6A), whereas those from monogenetic BVF cinder cones display a more narrow range toward the high end of the An spectrum (Fig. 6A). The mean An content of plagioclase phenocryst cores from BVF cinder cones is substantially higher (mean $\text{An}_{74.9}$) than those from VF stratovolcanoes (mean $\text{An}_{64.5}$) or OBVF central volcanoes (mean $\text{An}_{70.8}$). The same pattern of An variation occurs amongst plagioclase rim compositions (Fig. 6B).

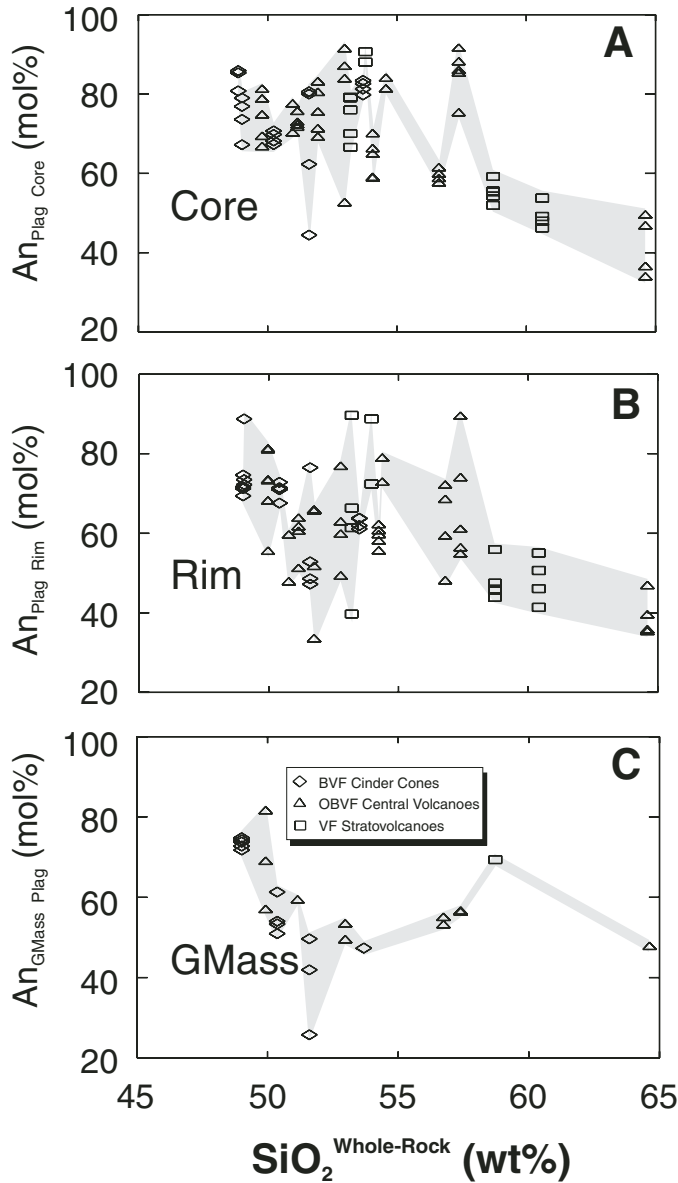


Figure 6. An content variations in plagioclase crystals versus whole-rock SiO_2 compositions. (A) Core An contents of plagioclase phenocrysts. (B) Rim An contents of plagioclase phenocrysts. (C) An contents of groundmass (GMass) plagioclase. Shaded compositional envelopes show no relationship between An content and host rock composition.

Oxygen Isotopic Compositions

The whole-rock oxygen isotope data from southeastern Guatemala span a wide range for young, fresh arc lavas (6.16–8.46‰). Basalts associated with subduction at convergent margins show enrichments in whole-rock $\delta^{18}\text{O}$ relative to mid-oceanic-ridge basalt (MORB) ($\delta^{18}\text{O} = +5.7\text{‰}$). Forearc trough and backarc basin basalts have a mean $\delta^{18}\text{O}$ of 5.9‰; oceanic arc basalts have a mean $\delta^{18}\text{O}$ of 6.0‰; and continental arc basalts average 6.2‰

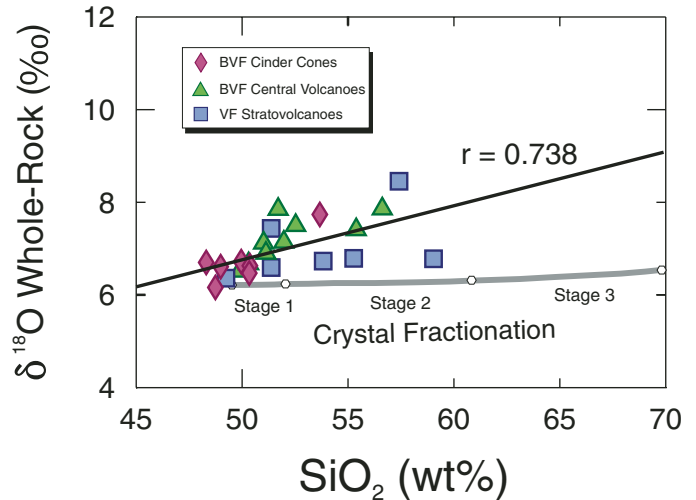


Figure 7. $\delta^{18}\text{O}_{\text{whole-rock}}$ versus SiO_2 diagram. $\delta^{18}\text{O}$ correlated positively with SiO_2 , with a Spearman rank correlation coefficient of +0.738. The relatively steep positive trend of the Guatemalan arc lavas contrasts with the near-horizontal crystal fractionation trend. Owing to the large range of compositions found at subduction zone volcanoes (i.e., from rhyolite to basalt), the crystal fractionation trend was modeled in three different stages. Stage 1, 2, and 3 refers to least squares mixing crystal fractionation models. See text for details. Behind the volcanic front (BVF) cinder cone lavas have lower $\delta^{18}\text{O}$ values than both volcanic front (VF) stratovolcanoes and older behind the volcanic front (OBVF) central volcanoes.

(Harmon and Hoefs 1995). Oxygen isotope analyses of olivines in oceanic arc lavas by laser fluorination are significantly less variable than the whole-rock data (Eiler et al., 2000) and might highlight the role of subsolidus alteration on whole-rock samples. The progressive enrichment in whole-rock $\delta^{18}\text{O}$ in arc settings may reflect interaction of parental magmas with volcanic arc basement and continental crust. Notably, $\delta^{18}\text{O}$ values in the arc lavas from southeastern Guatemala correlate positively with SiO_2 so that the most evolved lavas have the highest $\delta^{18}\text{O}$ values (Fig. 7). Among the three different edifice types in southeastern Guatemala, BVF cinder cones exhibit the best correlation ($r = 0.85$) and fall closest to MORB in oxygen isotopic composition. The mean $\delta^{18}\text{O}$ of the BVF cinder cones (+6.8‰) still exceeds the average oxygen isotopic composition of continental arc basalts. VF stratovolcanoes display a relatively poor correlation ($r = 0.41$), and the subhorizontal trend roughly parallels the crystal fractionation trend (Fig. 7). Lava from VF stratovolcanoes have a mean $\delta^{18}\text{O}$ value of +7.0‰, whereas the OBVF central volcanoes have a higher mean at $\delta^{18}\text{O} = +7.2\text{‰}$. A quartzite crustal sample has the highest $\delta^{18}\text{O}$ value measured in this study at 14.27‰. The granodiorite crustal sample has a $\delta^{18}\text{O}$ value of 3.71‰, suggesting alteration by a high temperature, meteoric fluid.

Light Element Contents

Boron (B) ranges from 1.1 to 22.9 ppm in the arc lavas from southeast Guatemala. The twenty-fold increase in abundance of

B underscores its highly incompatible chemical character. These analyses extend the narrow range of B concentration (2.4–15.2 ppm) found in an along-arc study of Central America (Leeman et al., 1994). The highest B values were recorded in more evolved lavas, and B contents correlate positively with SiO_2 (Fig. 8A).

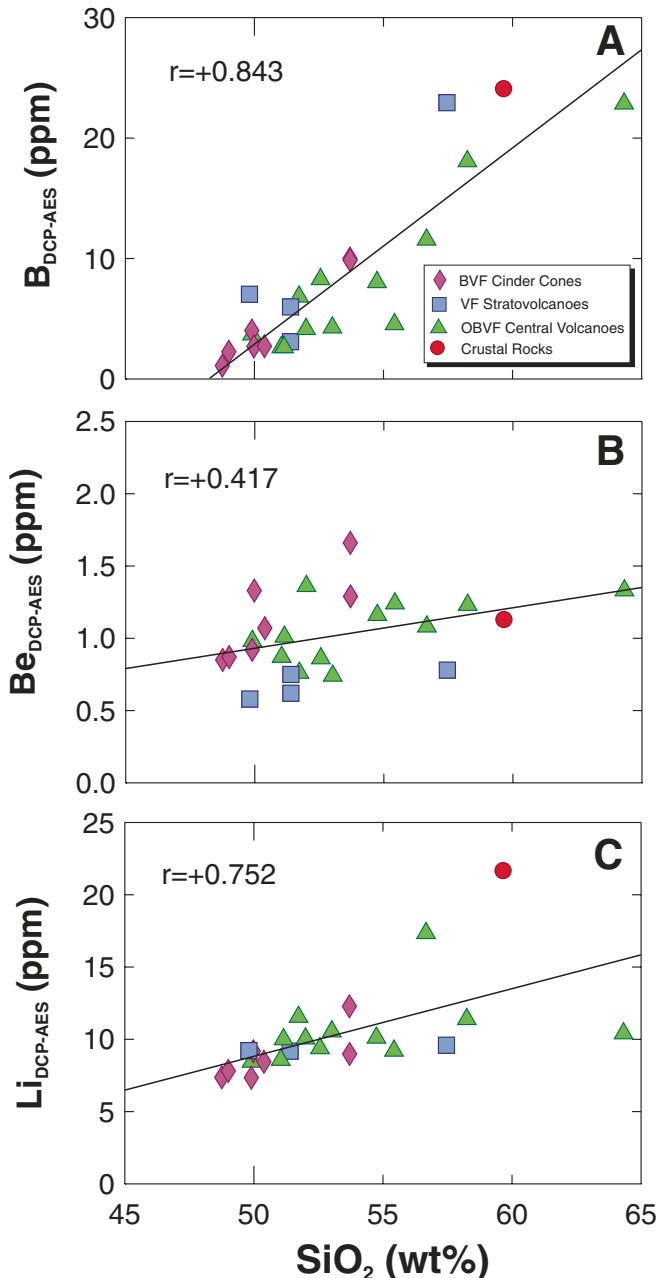


Figure 8. B, Be, and Li concentrations versus SiO_2 . The positive correlations with SiO_2 reflect the incompatible nature of the light elements. (A) B. (B) Be. (C) Li. Only the granodiorite crustal sample is plotted to represent a possible contaminant because quartzite plots far off the SiO_2 scale. BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front; DCP-AES—direct current plasma-atomic emission spectrometer.

VF stratovolcanoes and OBVF central volcanoes generally have higher B contents than the BVF cinder cones. Indeed, at a given SiO_2 content, VF stratovolcanoes tend to have slightly elevated B abundances relative to the central volcanoes and cinder cones. Granodiorite and quartzite samples taken to represent possible crustal contaminants contain B abundances comparable to the most enriched lavas (mean B of 23.1 ppm).

The 23 analyses reported here represent the first comprehensive study of Be in Guatemala. DCP-AES analyses of Be contents in rocks of this study display a limited range compared to B (0.6–1.7 ppm). The nearly threefold increase in Be affirms its less incompatible chemical character. Four other analyses record a limited range in Be of 0.5–0.8 ppm (Morris et al., 1990; Patino, 1997). In contrast to the behavior of B, the arc lavas exhibit a poor positive correlation between Be and SiO_2 and less systematic variation with respect to edifice type (Fig. 8B). Lavas from VF stratovolcanoes display the lowest Be contents, whereas BVF cinder cones and OBVF central volcanoes have a similar range of Be abundances and overall higher concentrations. At a given SiO_2 content, VF stratovolcanoes have lower Be abundances compared to the other edifice types. The two crustal samples analyzed cover a range of Be values (0.4–1.1 ppm) comparable to the lavas, with the granodiorite more enriched than the quartzite.

The Li analyses in this study represent the first compositional data of their type for arc lavas from Guatemala. Li contents analyzed by DCP-AES range between 7.3 and 17.3 ppm. The approximate twofold increase in Li concentrations suggests an incompatibility sequence of $\text{B} > \text{Be} > \text{Li}$. As for B and Be, a positive correlation exists between Li and SiO_2 ($r = +0.75$; Fig. 8C). BVF cinder cones generally have the lowest Li contents except for a sample far behind the volcanic front (GUC-25). VF stratovolcanoes exhibit uniform Li abundances of ~10 ppm and OBVF central volcanoes display the greatest variation from 8.4 to 17.3 ppm Li. The crustal samples span a larger range of Li composition than all the lavas with the granodiorite far exceeding the quartzite (21.7 versus 1.9 ppm Li).

DISCUSSION

Equilibrium in Minerals

Relations between bulk rock and phenocryst compositions provide a key perspective on whether a magma evolved in an open or closed system (Sakuyama, 1981; Hunter and Blake, 1995). Variations between mineral compositions and whole-rock compositions and assessment of whether equilibrium existed between mineral-mineral and mineral-whole-rock compositions lend insight into petrologic processes that influenced magma chemistry at depth. Obviously, these Guatemalan samples do not represent a cogenetic suite. The samples were collected from numerous locations at diverse edifice types. Indeed, the monogenetic nature of cinder cones essentially dictates sampling a single flow. Accordingly, this section places emphasis on the equilibrium approach. In fact, one might not expect systematic variation

in mineral chemistry with whole-rock compositions that do not represent a cogenetic suite. This caveat aside, the smooth major element trends suggest broadly common petrogenetic processes at Guatemalan volcanoes (Cameron et al., 2003).

Closed system evolution of a single magma chamber favors smooth systematic relationships between mineral and whole-rock chemistry (Davidson and de Silva, 1995; Hunter and Blake, 1995). However, the Guatemalan olivine phenocrysts show a range of compositions from Fe_{86} to Fe_{46} with no clear systematic variation with host rock composition (Figs. 3A and 3B). Groundmass olivine crystals exhibit similar compositional trends to the phenocryst rims (Fig. 3C). Compositional envelopes were drawn in Figure 3 to visually aid recognition of abrupt mineral chemical variations with changing whole-rock compositions. Mg numbers in clinopyroxene show an even more saw-toothed pattern of mineral variation with whole-rock composition (Fig. 4). A clinopyroxene phenocryst core from the most evolved sample analyzed records the maximum Mg number (Fig. 4A). Akin to clinopyroxene, Mg-rich orthopyroxene phenocrysts occur throughout the entire range of whole-rock compositions (Figs. 5A and 5B). Plagioclase phenocrysts exhibit many excursions to more anorthitic compositions at the basalt-basaltic andesite end of the whole-rock compositional spectrum but shift to a smooth declining trend at the andesite-dacite end (Figs. 6A and 6B). In summary, the overall lack of correlation between mineral chemistry and host lava composition hints strongly at the operation of open system processes. The existence of both normally and reversely zoned minerals in the same sample of some of the more evolved rocks suggests that magma mixing may be an important petrologic process. The lack of detailed compositional traverses across phenocryst phases prevents an extensive evaluation of this important petrologic process.

A more rigorous attempt to evaluate open system processes involves assessment of the degree of equilibrium attained by minerals with their host lavas. Under closed system fractional crystallization, equilibrium should exist between phenocryst cores and the whole-rock composition, between the groundmass melt and groundmass mineral phases, and between contiguous phenocrysts (Hunter and Blake, 1995). Consequently, plots of elemental ratios in minerals against identical ratios for bulk rocks should yield trends consistent with experimentally determined distribution coefficients.

Olivine-liquid K_D^{Fe-Mg} values (where $K_D^{Fe-Mg} = (FeO/MgO)_{olivine} / (FeO/MgO)_{whole-rock}$) decrease with increasing alkali content of coexisting liquid (Falloon et al., 1997). Based on the experiments of Falloon et al. (1997), a K_D^{Fe-Mg} value of 0.32 would characterize a basaltic melt with an alkali content ($Na_2O + K_2O$) of ~5 wt%. This value falls in the intermediate range determined by a series of experiments run by Sisson and Grove (1993). Core compositions of olivine phenocrysts scatter well off the equilibrium line constructed using a K_D^{Fe-Mg} of 0.32 (Fig. 9). A majority of the microprobe analyses plot above the equilibrium line in the field of olivine crystals too evolved for the whole-rock composition. Visual inspection of the data suggests that olivines from VF

composite cones and OBVF central volcanoes stray further from the equilibrium line than olivine from BVF cinder cones.

An equilibrium K_D^{Fe-Mg} value of 0.27 for orthopyroxene was determined by Gerlach and Grove (1982). In experimental melts, clinopyroxene has a slightly lower K_D^{Fe-Mg} value of 0.23 (Grove and Bryan, 1983). Electron microprobe analyses of orthopyroxene phenocryst cores plot predominantly in the upper field of crystals too evolved for the whole-rock composition (Fig. 10A). The VF stratovolcano and OBVF central volcano orthopyroxene grains deviate equally from the equilibrium line into the upper field (Fig. 10A); no orthopyroxene phenocrysts were measured in the BVF cinder cones. At higher $FeO/MgO_{whole-rock}$ compositions, the orthopyroxene cores extend to higher $FeO/MgO_{orthopyroxene}$ contents. The core compositions of clinopyroxene phenocrysts plot in the upper evolved crystal field, although more data points fall below the equilibrium line than for orthopyroxene (Fig. 10B). Clinopyroxene cores from OBVF central volcanoes exhibit the largest deviations below the equilibrium line, whereas OBVF central volcano and VF stratovolcano clinopyroxene cores show equal departures above the equilibrium line over the entire range of $FeO/MgO_{whole-rock}$. One core analysis of a clinopyroxene microphenocryst from a BVF cinder cone lava plots far into the field for evolved crystals.

Melting experiments using basalts and basaltic andesites suggest that K_D^{Ca-Na} values for plagioclase-melt equilibrium vary depending on the water content of the liquid (Sisson and Grove, 1993). Although independent of pressure, the K_D^{Ca-Na} value increases from 1.1 under anhydrous conditions to 5.5 within H_2O -saturated magmas at 2 kbar (Sisson and Grove,

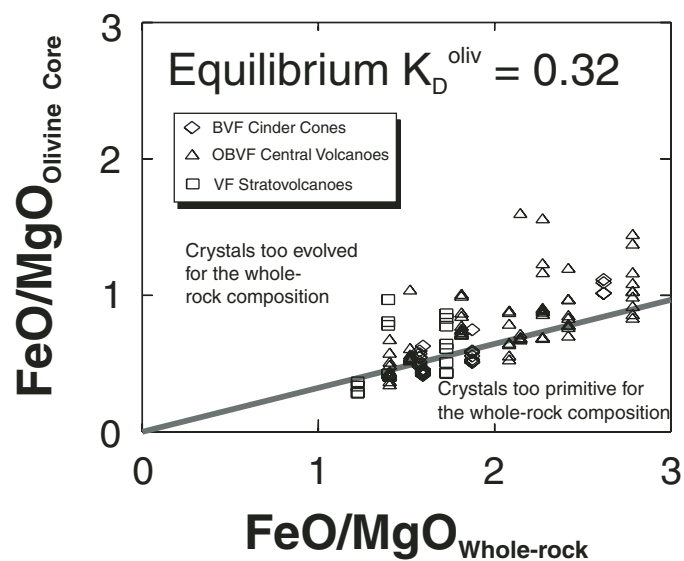


Figure 9. $FeO/MgO_{olivine}$ versus $FeO/MgO_{whole-rock}$. Core compositions of olivine phenocrysts largely fall above the equilibrium line ($K_D = 0.32$). BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

1993). Variable compositions of the melt created by magmatic differentiation exert no control on the K_D^{Ca-Na} value at constant water contents. Without an idea of the water content of the melt, interpreting the plagioclase equilibrium plot becomes a serious challenge. Water contents in melt inclusions hosted by olivines from both BVF cinder cones and VF stratovolcanoes are ~2.0 wt% (Walker et al., 2003). However, regardless of any reasonable water content and associated K_D^{Ca-Na} value chosen, the plagioclase core compositional data exhibit disequilibrium. Isolated VF stratovolcano and OBVF central volcano samples plot well above the equilibrium lines in the primitive crystal field (Fig. 11). Plagioclase from BVF cinder cones show smaller deviations from the equilibrium lines, with more uniform intermediate compositions, suggestive of moderate water contents in agreement with the melt inclusion data.

In a very general sense, any mineral core that falls in the evolved crystal field almost certainly did not crystallize from a liquid similar in composition to the bulk rock. Instead, these minerals likely formed from a more evolved liquid. Geologic explanations for the mineral cores that are too primitive for the whole-rock composition include early-formed phases that failed to react with the surrounding melt or crystals inherited from a more primitive liquid during magma mixing. These latter types definitely constitute the minority in this study.

Owing to the lack of pyroxene phenocrysts in the BVF cinder cones and the uncertainty in melt water contents with respect to plagioclase, we use olivine compositional data to try to determine whether closed or open system behavior was taking place. An argument for closed system behavior can be made if the electron microprobe did not analyze true olivine cores, but instead zones rimward of these points. True cores would be strictly defined as the first solid crystallizing from the bulk liquid. Thin sections are two-dimensional slices through a three-dimensional crystal, so there is no guarantee that the true core is analyzed when the electron beam is positioned at the center of a phenocryst. Thus, the numerous analyses that plot above the equilibrium line in the evolved field represent parts of the mineral that crystallized from interstitial liquid slightly more evolved than the bulk rock composition but still under equilibrium conditions. Although there is a coherent positive correlation between minimum FeO/MgO olivine core compositions with whole-rock FeO/MgO, the lack of true core compositions relative to rims plotting on the equilibrium line is troublesome.

If deviations from the mineral equilibrium line do indeed reflect disequilibrium and open system processes in the magma chamber, then the magnitude of the deviation logically holds important information on the extent of the process. It becomes a trivial matter to calculate the mean deviation from the olivine equilibrium line for each volcano type. Olivine cores from BVF cinder cones have a range in deviations from the equilibrium line of -0.12 to +0.29 with a mean of 0.01. OBVF central volcanoes have a range of deviations from -0.14 to +0.90 with a mean of 0.10, and VF stratovolcanoes exhibit a range of -0.11 to +0.56 with a mean deviation of 0.12. Based on the magnitudes of these

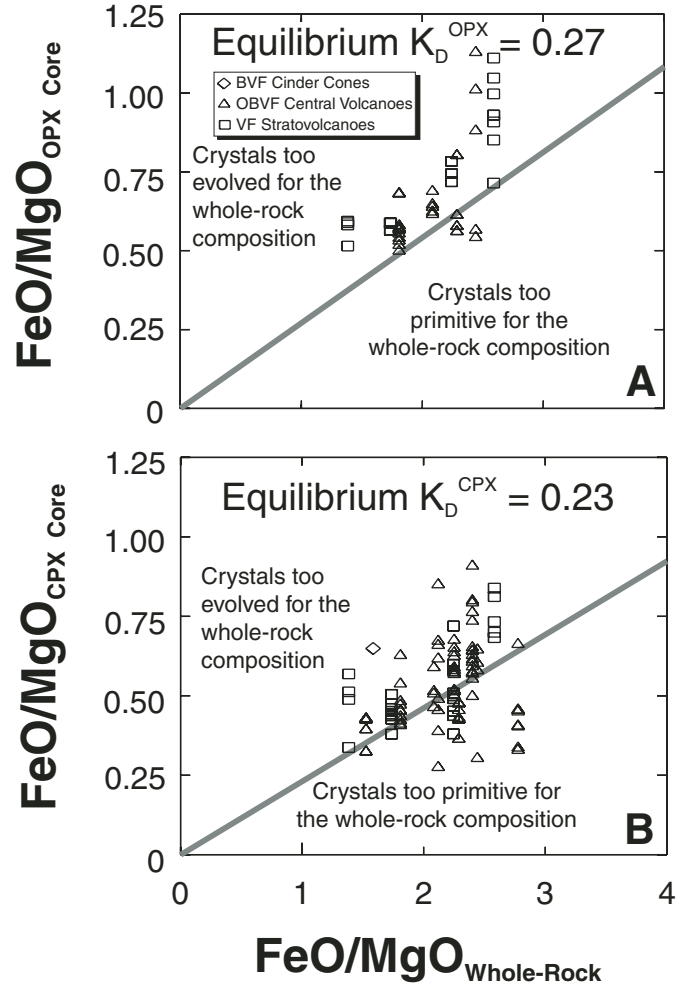


Figure 10. FeO/MgO_{pyroxene} versus FeO/MgO_{whole-rock} diagrams. (A) Orthopyroxene (OPX) phenocryst core data plot largely above an equilibrium defined by $K_D = 0.27$. (B) Clinopyroxene (CPX) phenocryst core data also plot largely above the equilibrium line defined by $K_D = 0.23$. BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

deviations, the open system process seems more prevalent at the VF stratovolcanoes and OBVF central volcanoes.

Assuming that true core regions of phenocrysts were analyzed more often than not by the electron microprobe, the non-systematic variation of mineral chemistry with whole-rock composition and the nonequilibrium mineral core compositions together lead to the conclusion that open system processes played a significant role in creating compositional diversity in the Guatemalan lavas.

Oxygen Isotope Variations

Two features of the whole-rock $\delta^{18}O$ data warrant comment. First, the arc lavas of southeastern Guatemala have elevated values of $\delta^{18}O$ compared to their respective global counterparts. The

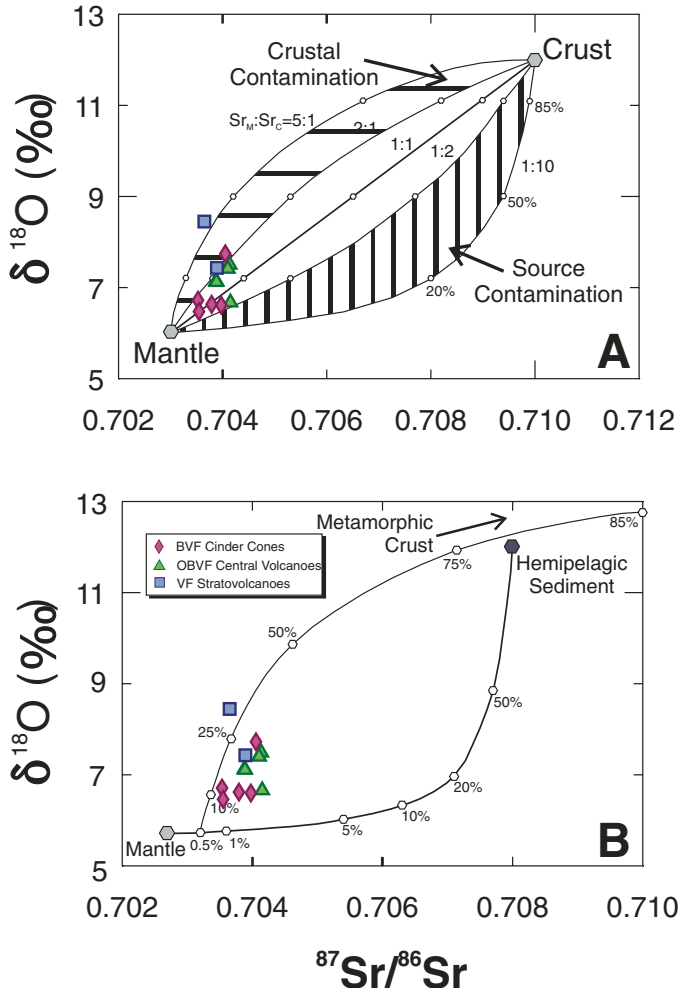


Figure 11. $\text{CaO}/\text{Na}_2\text{O}_{\text{plagioclase}}$ versus $\text{CaO}/\text{Na}_2\text{O}_{\text{whole-rock}}$ diagram. The equilibrium K_D value depends on the water content of the magma. K_D values of 1.1, 3.4, and 5.5 are shown. The scatter exhibited by the plagioclase core data suggests disequilibrium behavior. BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

range in BVF cinder cones from 6.16 to 7.74‰ falls above the global mean for whole-rock backarc basins of 6.0‰ (Harmon and Hoefs, 1995). Likewise, the range for VF stratovolcanoes of 6.36–8.46‰ lies above the mean for continental arc basalts of 6.2‰ (Harmon and Hoefs, 1995). Second, relatively large ranges in $\delta^{18}\text{O}$ characterize the basalts and evolved lavas from southeastern Guatemala.

Crystal Fractionation

The positive correlation between $\delta^{18}\text{O}$ and SiO_2 might suggest that crystal fractionation played an important role in the chemical variation (Fig. 7). However, the small isotopic fractionations that occur between melt and crystals at high temperature suggest that crystal differentiation has little influence in changing the $\delta^{18}\text{O}$ of evolving magmas. Oxygen isotope studies of closed system

magmatic suites confirm this general observation (Muehlenbachs and Byerly, 1982; Chivas et al., 1982; Kalamarides, 1984). At the Galapagos Spreading Center, 90% fractionation from low K-tholeiite to rhyodacite only enriched the residual melt by $\sim 1.2\text{‰}$ (Muehlenbachs and Byerly, 1982). Crystal fractionation in the Koloula Igneous Complex on the island of Guadalcanal generated $\delta^{18}\text{O}$ values that range from 5.7 to 7.2‰ (Chivas et al., 1982). $\delta^{18}\text{O}$ values of the Kiglapait layered intrusion vary from 6.0‰ for the first liquids in the Lower Zone to 6.3‰ for the final liquids in the Upper Zone (Kalamarides, 1984). Thus, empirical knowledge suggests that the range of $\delta^{18}\text{O}$ values in the Guatemalan arc lavas exceeds the normal fractionation spread. In fractionating suites of magmas with demonstrable closed system behavior, the slope of the $\delta^{18}\text{O}$ versus SiO_2 diagram varies according to temperature.

Further insight on the $\delta^{18}\text{O}$ variations can be gleaned from quantitative modeling (Sheppard and Harris, 1985; Gruner, 1987; Woodhead et al., 1987; Harmon and Gerbe, 1992; Singer et al., 1992). Closed system Rayleigh fractional crystallization enriches the residual liquid in $\delta^{18}\text{O}$ according to the equation:

$$\delta^{18}\text{O}_{\text{melt}}^t = (\delta^{18}\text{O}_{\text{melt}}^i + 1000) f^{\alpha-1} - 1000, \quad (1)$$

where $\delta^{18}\text{O}_{\text{melt}}^t$ is the $^{18}\text{O}/^{16}\text{O}$ ratio of the magma after some finite amount of crystal fractionation, $\delta^{18}\text{O}_{\text{melt}}^i$ is the initial $^{18}\text{O}/^{16}\text{O}$ ratio of the magma prior to fractionation processes, f is the fraction of melt remaining at time t , and α is the mean crystal-melt fractionation factor. In high temperature magmatic systems, α values approximate unity, so the relative enrichment of the crystal-melt phases in ^{18}O is given simply by

$$\Delta_{A-B} (\text{‰}) = 10^3 \ln \alpha_{A-B} = \delta_A - \delta_B. \quad (2)$$

The mean fractionation factor can be uniquely calculated for distinct stages in the evolutionary history of the Guatemalan magmas. The calculation involves multiplying the molar proportion of each phase by the mole fraction of oxygen in each phase and the mean $\delta^{18}\text{O}$ fractionations between crystal and melts taken from the literature (e.g., Sheppard and Harris, 1985; Kalamarides, 1986). Substitution of the mean fractionation factor into the general Rayleigh fractionation formula allows calculation of the expected $\delta^{18}\text{O}$ value of residual melts.

Table 5 provides the details of least squares crystal fractionation model developed to support interpretation of the oxygen isotope data. In Stage 1, 28.6% crystallization of plagioclase, clinopyroxene, and olivine from parent magma GUC-800 (recalculated $\text{SiO}_2 = 49.53\%$) yields the daughter magma GCH-01 (recalculated $\text{SiO}_2 = 52.04\%$) with an acceptable $r^2 = 0.805$. The mean fractionation factor ($\Delta_{\text{xl-melt}}$) is -0.06 or $\alpha = 0.99994$. The calculated $\delta^{18}\text{O}_{\text{melt}}$ after 28.6% crystallization using a $\delta^{18}\text{O}_{\text{melt}}^i$ of 6.21‰ is 6.23‰.

Plagioclase, clinopyroxene, olivine, and titanomagnetite could crystallize from GCH-01 in Stage 2 to produce a nearly equal volume of daughter melt similar in composition to FEL-3 (recalculated $\text{SiO}_2 = 60.83\%$) with $r^2 = 0.880$. The resulting Δ_{xl}

TABLE 5. LEAST SQUARES MIXING CRYSTAL FRACTIONATION MODEL

| Stage | Parent | Proportion | Percentage | Mineral | Fit |
|-------|---------|------------|------------|---|---------------|
| 1 | GUC-800 | 0.123 | 43.30% | Plagioclase (An ₉₂) | $r^2 = 0.805$ |
| | | 0.12 | 42.30% | Cpx (Wo ₄₆ -En ₃₇ -Fs ₁₇) | |
| | | 0.041 | 14.30% | Olivine (Fo ₇₂) | |
| | | 0.714 | | GCH-01 (daughter) | |
| 2 | GCH-01 | 0.029 | 6.00% | Ti-magnetite (Ti ₁₀) | $r^2 = 0.880$ |
| | | 0.263 | 56.00% | Plagioclase (An ₈₇) | |
| | | 0.085 | 18.00% | Cpx (Wo ₄₄ -En ₄₇ -Fs ₉) | |
| | | 0.093 | 19.80% | Olivine (Fo ₆₈) | |
| | | 0.517 | | FEL-3 (daughter) | |
| 3 | FEL-3 | 0.04 | 11.00% | Ti-magnetite (Ti ₁₂) | $r^2 = 0.095$ |
| | | 0.256 | 71.00% | Plagioclase (An ₅₉) | |
| | | 0.023 | 6.30% | Cpx (Wo ₄₂ -En ₄₄ -Fs ₁₄) | |
| | | 0.042 | 11.70% | Olivine (Fo ₆₀) | |
| | | 0.638 | | GAG-13A (daughter) | |

is -0.121 or $\alpha = 0.99988$. The Raleigh fractionation formula yields a calculated $\delta^{18}\text{O}$ of 6.31‰ for the melt.

In Stage 3, 46.2% crystallization of plagioclase, clinopyroxene, olivine, and titanomagnetite from a FEL-3 parental melt produces a daughter magma GAG-13A (recalculated $\text{SiO}_2 = 69.81\%$) with a very low r^2 of 0.095. More significant titanomagnetite fractionation increased the mean fractionation factor (Δ) to -0.492 or $\alpha = 0.99951$. The calculated $\delta^{18}\text{O}_{\text{melt}}$ at the end of the fractionating sequence is 6.53‰ . Large shifts in $\delta^{18}\text{O}$ of magma (i.e., $>1\text{‰}$) result only through the removal of significant amounts of either an ^{18}O -depleted phase such as olivine or an oxide mineral or an ^{18}O -enriched phase such as quartz (Harmon and Gerbe, 1992). Otherwise, closed system differentiation of magma from a basalt to rhyolite generates much smaller variations in $\delta^{18}\text{O}$. The predominance of plagioclase in the fractionating assemblage of calc-alkaline magmatic systems often results in $\delta^{18}\text{O}$ changes of <0.5 . Table 6 presents a summary of the calculated variations in $\delta^{18}\text{O}$ of a melt dictated by crystal fractionation only. The oxygen isotope variations traced through the three stages define a near-horizontal crystal fractionation line on the $\delta^{18}\text{O}$ versus SiO_2 diagram (Fig. 7). Open-system processes produce deviant $\delta^{18}\text{O}$ - SiO_2 arrays compared with the near-horizontal crystal fractionation trend when the contaminant has an isotopic signature distinct from the melt (Harmon and Gerbe, 1992). Mafic to intermediate magmas that assimilate ^{18}O -rich crust during differentiation form steep, positive $\delta^{18}\text{O}$ - SiO_2 trends (Davidson and Harmon, 1989; Ellam and Harmon, 1990),

whereas assimilation of ^{18}O -depleted crustal rocks modified by meteoric hydrothermal alteration yield negative $\delta^{18}\text{O}$ - SiO_2 arrays (Grunder, 1987; Harmon and Gerbe, 1992). The positive array with a Spearman rank correlation coefficient of $+0.738$ defined by the Guatemalan arc lavas (Fig. 7) favors involvement of an enriched ^{18}O contaminant. The groundmass of lava flows readily exchange oxygen during low temperature weathering or interaction with hydrothermal fluids, and conceivably could account for some of the positive shift in the $\delta^{18}\text{O}$ data. The random nature of alteration argues against it being a significant variable in light of the respectable correlation between $\delta^{18}\text{O}$ and SiO_2 . Trace element mobility has been documented in spheroidally weathered Guatemalan lavas (Patino et al., 2003), but this study avoided visibly weathered samples and collected only the freshest samples from the volcanoes in question. Ideally, complementary measurements of the oxygen isotopic composition of refractory minerals such as olivine or clinopyroxene by laser fluorination or ion microprobe techniques would more accurately determine magmatic values (Singer et al., 1992; Macpherson and Matthey, 1998; Eiler et al., 1998, 2000; Dorendorf et al., 2000; Vroon et al., 2001). Correction procedures based on assumptions of primary H_2O contents generally result in small adjustments of $\delta^{18}\text{O}$ values. Lavas from Martinique in the Lesser Antilles typically had measured $\delta^{18}\text{O}$ reduced by $<0.5\text{‰}$ (Davidson and Harmon, 1989). Owing to the young age and fresh nature of the Guatemalan lavas in thin section and the small modifications attributed to water uptake, crustal contamination plays a more prominent role in enriching the magmas in $\delta^{18}\text{O}$.

Source versus Crustal Contamination

Oxygen isotopic compositions in combination with radiogenic isotopes can distinguish, in principle, between crustal and source contamination (James, 1981). Sr/O ratios of the mantle and crustal end members define mixing trajectories on a plot of $\delta^{18}\text{O}$ versus $^{87}\text{Sr}/^{86}\text{Sr}$ (Fig. 12). The high Sr content of mafic magmas compared to that of the potential crustal contaminants accounts for a convex crustal contamination mixing curve.

TABLE 6. CALCULATED VARIATIONS IN $\delta^{18}\text{O}$ OF A MELT CONTROLLED BY CRYSTAL FRACTIONATION

| Model | $\delta^{18}\text{O}$ Values | | Mean Δ (‰) | $\Delta = \delta^{18}\text{O}_{\text{xt}} - \delta^{18}\text{O}_{\text{magma}}$ | | | |
|---------|------------------------------|----------|----------------------|---|------|------|------|
| | Parent | Daughter | | Plag | Cpx | Oliv | Mt |
| Stage 1 | 6.21 | 6.23 | -0.06 | 0.2 | -0.2 | -0.5 | 0 |
| Stage 2 | 6.23 | 6.31 | -0.121 | 0.2 | -0.2 | -0.5 | -1.9 |
| Stage 3 | 6.31 | 6.53 | -0.492 | 0 | -1 | -1.8 | -3.6 |

Conversely, the low Sr content of the bulk mantle relative to subducted crustal material accounts for a concave source contamination trend. Partial melting of mantle peridotite produces primary magma enriched in incompatible elements such as Sr and Nd relative to lithologies of the continental crust. In sharp contrast, the bulk mantle has low incompatible element contents relative to enriched sediments and thus represents the depleted end member in the source contamination case. These differing chemical characteristics produce convex trajectories for crustally contaminated arc magmas and concave trends after recycling sediments into a MORB-like subarc mantle source.

Whole-rock oxygen isotopic data have poor but positive correlations with the Sr isotopic composition of the Guatemalan arc lavas. Mixing curves can be calculated using the binary mixing equation:

$$R_M^X = [R_A^X X_A f + R_B^X X_B (1 - f)] / [X_A f + X_B (1 - f)], \quad (3)$$

where R_M^X is the isotope ratio of Sr or O in a mixture of components A and B, X_A and X_B are concentrations of Sr or O in A and B, and f is the weight fraction of A. Each curve was calculated using a different ratio of Sr in the mantle versus Sr in the crust (i.e., $Sr_M:Sr_C$). Five different concentrations (100, 40, 20, 10, and 2 ppm) were assumed for the mantle, whereas the enriched end member was taken to have a Sr concentration of 20 ppm. The model calculations assume equal concentrations of O in the two end members but different $\delta^{18}O$ values of 6.0‰ for the mantle and 12.0‰ for the crust. Varying values of f and substitution of these concentrations into the general two-component binary mixing equation generates the five separate curves. The data plot within or in close proximity to the field representing magmas contaminated by continental crust (Fig. 12A). The spread of the data suggests that between 5% and 35% crustal material is added to the parental arc magma.

A more detailed, two-stage model can be constructed based on the distribution of data in the simple model using more geochemically relevant parameters. Two-stage models have been successful in explaining Sr and O isotopic relationships in the Marianas (Woodhead et al., 1987) and the Aeolian Islands (Ellam and Harmon, 1990). Source contamination shifts the data away from the mantle source in the direction of higher $^{87}Sr/^{86}Sr$, whereas crustal contamination profoundly increases the $\delta^{18}O$ contents of the magma compared to only moderate increases in $^{87}Sr/^{86}Sr$. Again, oxygen concentrations were assumed to be constant in all components. The mantle Sr concentration of 15 ppm (Chen and Frey, 1985), $^{87}Sr/^{86}Sr$ ratio of 0.7027 (Zindler and Hart, 1986), and $\delta^{18}O$ of 5.7 (Woodhead et al., 1987) characterize a MORB source. The mean hemipelagic sediment from Deep Sea Drilling Project (DSDP) Site 495 off Guatemala has ~300 ppm Sr and a $^{87}Sr/^{86}Sr$ ratio of 0.708 (Patino, 1997). The sediments from DSDP 495 were not analyzed for $\delta^{18}O$. Unaltered hemipelagic sediment from Middle Valley in the northern Pacific Ocean has $\delta^{18}O$ values that approach 12.0‰ (Goodfellow et al., 1993). Extrapolation of the Guatemalan Sr-SiO₂ trend back to a SiO₂ content of

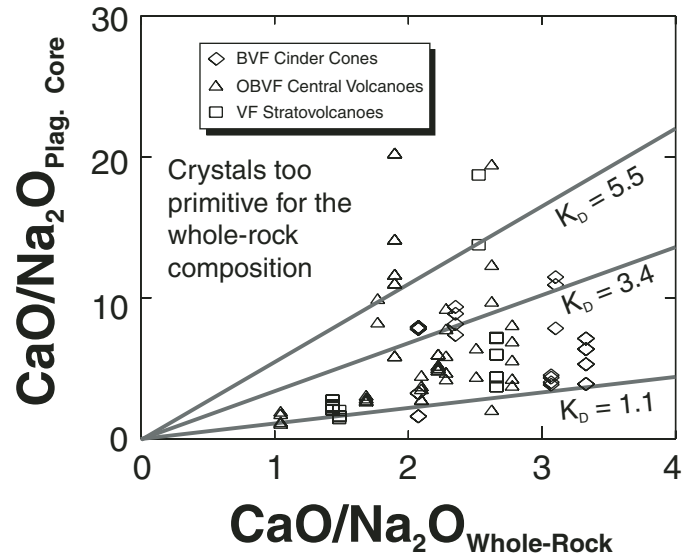


Figure 12. (A) $\delta^{18}O$ versus $^{87}Sr/^{86}Sr$ diagram (James, 1981). The Guatemalan arc lavas plot within the field of magmas contaminated by continental crust as opposed to mantle contaminated by sediments. (B) Diagram of $\delta^{18}O$ versus $^{87}Sr/^{86}Sr$ showing a two-stage evolution model for arc lavas from southeastern Guatemala. Crustal contamination of primary magmas by metamorphic crustal samples follows limited contamination of the mantle by hemipelagic sediments. A series of mixing hyperbolas between sediment-modified mantle and crustal contaminants would best fit the spread of data points. BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front.

46% approximates the composition of a hypothetical primitive arc magma in Guatemala and yields the model Sr abundance for the magma of 750 ppm. The magma $^{87}Sr/^{86}Sr$ ratio of 0.7032 and $\delta^{18}O$ value of 5.73 were products of the Stage 1 model calculation of 0.5% sediment contamination in the source. Finally, quartzite in southeastern Guatemala has Sr contents of ~30 ppm, a $^{87}Sr/^{86}Sr$ ratio of 0.740, and a $\delta^{18}O$ value of 14.0‰ (Walker et al., 1995; this study). The quartzite sample was incorporated into the model simply because of the altered nature and unrealistically low $\delta^{18}O$ value of 3.71‰ determined for the granodiorite sample. The high $\delta^{18}O$ value of the quartzite used in this model would approximate other more likely crustal contaminants.

The two-stage model illustrated in Figure 12B represents a more realistic explanation of the contamination history in Guatemala. A combination of between 0.5% and 2.0% hemipelagic sediment addition to a MORB-source mantle and subsequent mixing of magma with between 10% and 30% metamorphic crustal material could explain the Sr-O isotopic data. The proposed two-stage process conforms to the general conceptual model for arc magmatism in which material transfer at the slab-mantle wedge interface induces partial melting and ascending magmas digest crustal materials during stagnation in reservoirs at various depths.

Effects of Crustal Contamination on Elemental Concentrations

Plots of various elements versus $\delta^{18}\text{O}$ should convey information about elemental inputs to magmas during crustal contamination assuming that assimilation accounts for increases in $\delta^{18}\text{O}$. Two statistical quantities contribute valuable insight into the nature of the relationship between the two participating variables. The Spearman rank correlation coefficient r essentially measures how well the data fit a straight line. A high value of r represents a high degree of correlation. The threshold r value that indicates a statistically significant correlation depends on the number of samples. On the other hand, the slope of the linear regression line relates to the dependence or independence of one variable relative to the other. Simple examples illustrate their interpretative value. A data set with a high r value near one and a slope near zero suggests an excellent correlation, but also that y is independent of x . A second data set with $r = 0.75$ and $m = 1$ indicates a moderate to good correlation and that y depends on x , or in geochemical phrasing, y and x behave similarly. In the current discussion, $\delta^{18}\text{O}$ is used to evaluate the effect of crustal contamination on critical incompatible elements. Spearman rank correlation coefficients screen combinations for statistical significance, whereas slope measures the magnitude of the behavioral similarity.

Overall, incompatible element versus $\delta^{18}\text{O}$ plots feature relatively high Spearman rank correlation coefficients. The key threshold value for correlation coefficients with a 0.1% level of significance are 0.618 at $n = 23$ (LeMaitre, 1982). Be, Sb, and La have correlation coefficients below the threshold value and consequently appear unrelated to $\delta^{18}\text{O}$. Compilation of the slope data suggests that incompatible elements exhibit a complete spectrum of coherency with $\delta^{18}\text{O}$ directly and crustal contamination indirectly (Fig. 13). Crustal contamination more significantly affects those incompatible elements with a slope of ~ 1 or greater (i.e., B, La, Pb, Li, and Th). Assimilation of crust exerts less control on incompatible element concentrations of Rb, Sb, Ba, and Be based on slopes < 0.3 . An intermediate group of elements with slopes between 0.3 and 0.9 consists of Cs and U. Thus, the sequential order of the incompatible elements on the x -axis from left to right in Figure 13 serves as a relative measure of the chemical influence of crustal contamination.

Igneous petrologists rely heavily on incompatible trace element ratios to provide important petrogenetic information. Common differentiation processes such as crystal fractionation and partial melting have little effect on ratios involving trace elements with similar degrees of incompatibility. Thus, any information on the nature of elemental modification by crustal contamination helps petrologists make more intelligent decisions when assembling these key trace element ratios. Crustal contamination will influence a ratio formed from elements with disparate slopes (Fig. 13). B/Be ratios in Guatemalan arc lavas correlate positively with SiO_2 and trend toward but do not exceed the field of crustal rocks (Fig. 14A). Many studies of continental arc lavas indiscriminately use B/Be as a measure of the slab signature

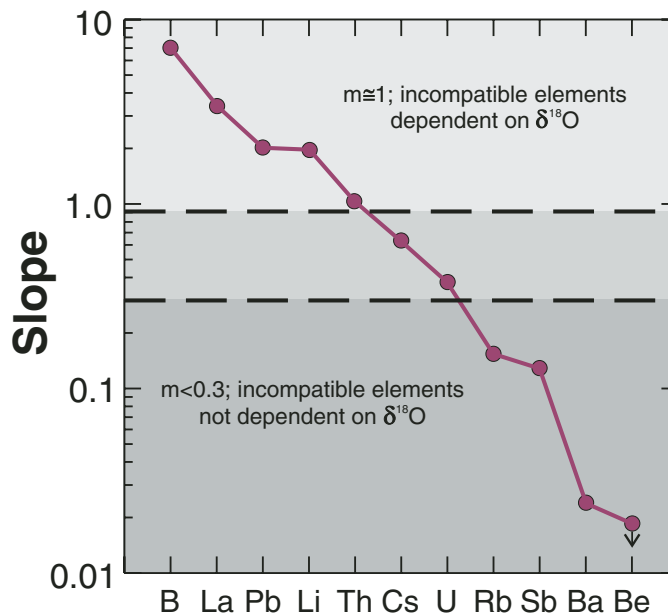


Figure 13. Magnitude of the slope for select trace elements and $\delta^{18}\text{O}$. Given that slope measures the interdependency of variables on variation diagrams, elements with slopes greater than 0.9 have been most affected by crustal contamination.

without thorough assessment of the control exercised by crustal contamination. The effects of crustal contamination will be canceled out in ratios involving elements with similar slopes. Not only does a poor correlation exist between Ba/La versus SiO_2 , but many of the lavas possess higher Ba/La ratios than the analyzed crustal rocks (Fig. 14B). Accordingly, Ba/La monitors slab signatures much better than a ratio such as B/Be in an arc with thick continental crust.

Edifice Style and Crustal Contamination

Porphyritic lavas with low-pressure phase assemblages develop at the large stratovolcanoes of the VF owing to extensive crystallization in shallow magma chambers. Analogous rocks with similar phenocryst assemblages occur at OBVF central volcanoes. In sharp contrast, the mafic BVF cinder cone lavas have aphyric textures and were produced by small volume eruptions associated with the extensional regime of the Ipala Graben. These diffuse vents often coincide with systematic lineaments interpreted as deep crustal structures. The whole-rock oxygen isotope data support this volcanological scenario. The highly elevated $\delta^{18}\text{O}$ values measured at the VF stratovolcanoes and OBVF central volcanoes result from crustal assimilation accompanying fractional crystallization in high-level magma chambers. Erupted lavas have common low-pressure phenocryst phases and evolved whole-rock compositions altered by the assimilation of crustal contaminants. The slightly elevated $\delta^{18}\text{O}$ values observed in the BVF cinder cone lavas indicate that the magmas ascended

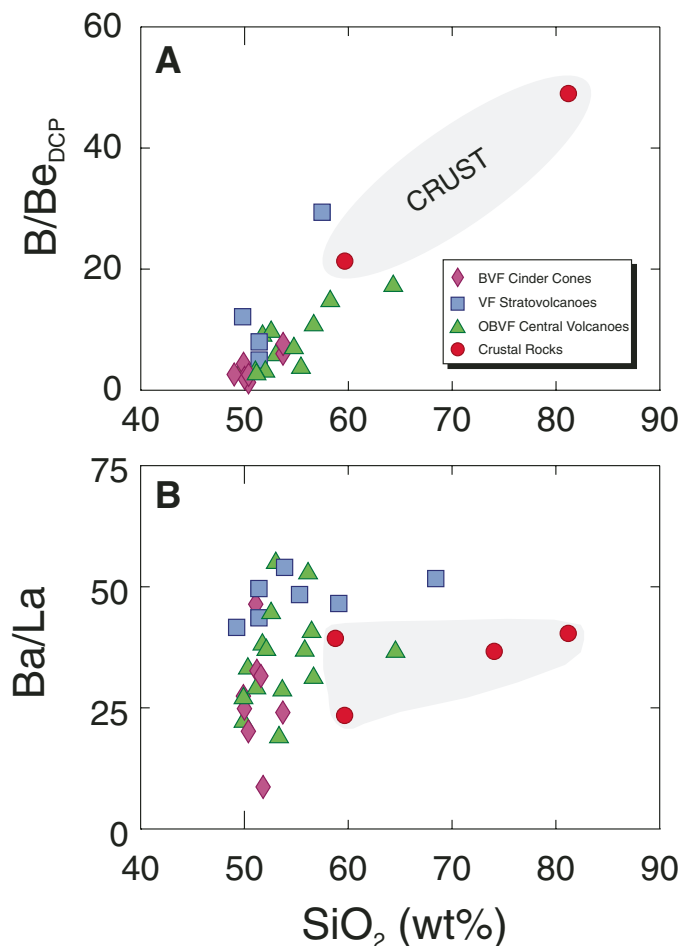


Figure 14. Critical trace element ratios versus SiO_2 . (A) B/Be versus SiO_2 . Guatemalan arc lavas essentially plot on a mixing line between mantle and crust. (B) Ba/La versus SiO_2 . Ba/La in Guatemalan arc lavas exceeds that recorded in possible crustal contaminants. Ba/La ratio was less affected by crustal contamination than B/Be . BVF—behind the volcanic front; OBVF—older behind the volcanic front; VF—volcanic front; DCP—direct current plasma.

more rapidly from deep crustal magma chambers underlying the extended crust. These eruptive products have more mafic compositions, an aphyric to sparsely olivine-phyric texture, and little chemical crustal signature. Short residence times in shallow reservoirs suggest that the small enrichment in $\delta^{18}\text{O}$ probably records deeper crustal contamination processes. Unusual trends on Pb isotope plots for the BVF cinder cones identified granulitic crust as a possible contaminant (Walker et al., 1995). Curiously, cinder cone lavas in Mexico seemed vulnerable to crustal assimilation based on their B/Be - SiO_2 relationships (Hochstaedter et al., 1996). Small batches of magma from monogenetic centers in Mexico traversed separate sections of crust, and therefore may be more susceptible to crustal contamination. A supplementary oxygen isotope study in the Mexican volcanic belt might help resolve this discrepancy.

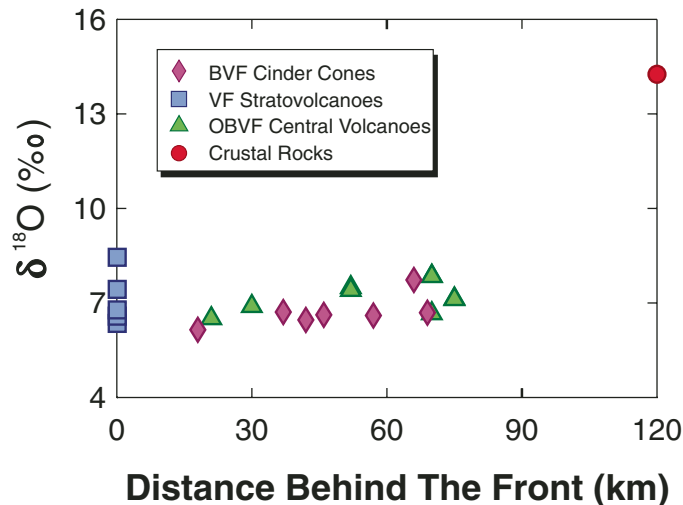


Figure 15. $\delta^{18}\text{O}$ versus distance behind the volcanic front in southeastern Guatemala. $\delta^{18}\text{O}$ remains essentially constant at behind the volcanic front (BVF) cinder cones across the arc, implying no absolute increase in crustal contamination. BVF cinder cones have $\delta^{18}\text{O}$ values only slightly above normal mid-oceanic ridge basalt. OBVF—older behind the volcanic front; VF—volcanic front.

Across-Arc Changes in $\delta^{18}\text{O}$

Nd and Pb isotopic ratios in BVF cinder cones display relatively consistent and continuous across-arc changes: both $^{207}\text{Pb}/^{204}\text{Pb}$ and $^{208}\text{Pb}/^{204}\text{Pb}$ generally increase, whereas $^{143}\text{Nd}/^{144}\text{Nd}$ declines with distance behind the front (Walker et al., 1995). Either an absolute or apparent increase in crustal contamination across the arc can explain the regular radiogenic isotopic variations. Walker et al. (1995) discounted an absolute increase in the amount of contamination as unreasonable owing to the likelihood of rapid magma ascent caused by tectonic extension behind the front. Instead, assimilation of older, more isotopically evolved crust farther behind the front only creates the illusion of increasing extents of crustal contamination. Oxygen-isotopic compositions do not evolve with age like the radiogenic isotopes, and therefore offer another means of evaluating this question. The $\delta^{18}\text{O}$ values of the BVF cinder cones remain relatively constant across the arc (Fig. 15), ruling out the possibility that the overall amount of contamination increases behind the front. The highest $\delta^{18}\text{O}$ values occur at VF stratovolcanoes and at OBVF central volcanoes.

CONCLUSIONS

Geochemical studies of Guatemalan arc lavas demonstrate the importance of open system processes in modifying mantle melts. Mineral chemistry varies nonsystematically with whole-rock compositions and together with disequilibrium core compositions in olivine, pyroxene, and plagioclase indicate that the

crustal magma chambers behave as an open system. Whole-rock oxygen isotope data are compatible with crustal contamination being an important open system process in southeastern Guatemala. Crystal fractionation cannot explain the large ranges in $\delta^{18}\text{O}$ measured in the erupted lavas. Rather, steep arrays on the $\delta^{18}\text{O}$ - SiO_2 diagram favor incorporation of ^{18}O -rich crustal rocks during ascent of primitive magmas. Enrichments in $\delta^{18}\text{O}$ imposed by low-temperature weathering or alteration likely prove insignificant compared to the effects of crustal contamination. Combined Sr and O isotope models indicate minor additions of 0.5%–1% sediment to the mantle wedge followed by variable degrees of crustal contamination between 10% and 30%. Lower $\delta^{18}\text{O}$ values from BVF cinder cones compared to the stratovolcanoes of the VF and OBVF suggest that relatively lower amounts of crustal contamination accompany smaller scale, deeper magmatic systems. Extension associated with the Ipala Graben promoted short crustal residence times for the mafic cinder cone magmas and contributed to the lower $\delta^{18}\text{O}$ values. Variations in $\delta^{18}\text{O}$ better reflect the magnitude of the crustal contamination process than the age-controlled radiogenic isotopes. Open system processes such as crustal contamination and magma mixing might control eruption style, leading to the construction of different edifice types. More significant volcanic hazards would therefore be associated with the eruption of VF stratovolcanoes and OBVF central volcanoes.

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